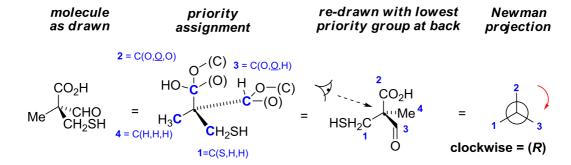
Stereochemistry 2011-2012 EXAM-STYLE QUESTION - OUTLINE ANSWERS

(a) Applying the CIP rules we obtain:



NB. 'ghost' atoms, as defined in the notes, are shown in brackets on the structure and underlined in the 'priority summation' (in blue)

Ph
$$(C) \stackrel{(C)}{H} \stackrel{(C)}{C} \stackrel{(C)}{$$

(b) Only one of the following molecules, molecule C, has an asymmetric enantiomeric form – see below. Molecule B is achiral due to an internal mirror plane (*i.e.* an S_I improper rotation axis) and molecule A has a C_2 axis and therefore has a dissymmetric enantiomeric form.

Me H H H
$$H$$

Me H H

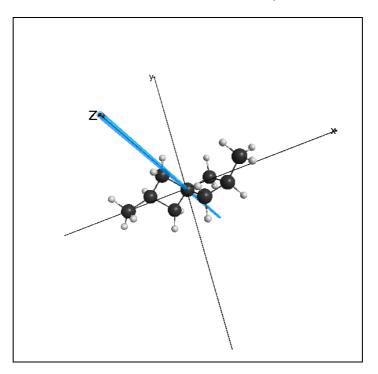
A H

Gissymmetric enantiomers $(C_s = S_1 \text{ 'axis'})$ (achiral)

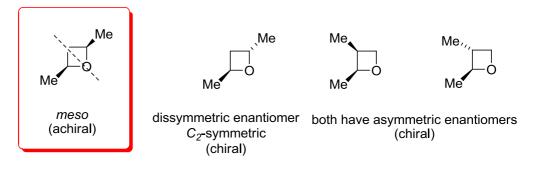
 $(C_s = S_1 \text{ 'axis'})$ (achiral)

It is non-trivial to spot that molecule **A** has a C_2 axis of symmetry. A 3D model of this compound has been uploaded here: http://www.ch.ic.ac.uk/spivey/teaching/org1stereochemistry/stereochemistryjm ols/0910stereochemistry5.html to allow you to rotate the molecule (A model of

molecule **B** is also there, showing its internal mirror plane). The below image was generated using the program 'wxMacMolPlt' which can be downloaded free from: http://www.scl.ameslab.gov/MacMolPlt/. This program has the facility to allow you to identify symmetry operations within any 3D molecule coordinates. In this case, the C_2 axis is coincident with the Z axis (as defined for the coordinates of molecule at the above URL).



(c) The cis-1,3-dimethyl compound is a meso compound as it is achiral by virtue of an internal mirror plane (S_I axis) but has a stereoisomer that is chiral: the trans-1,3-dimethyl compound, which has a dissymmetric enantiomeric form. The cis-and trans-1,2-dimethyl compounds both have asymmetric enantiomers.



Meso is the term given to achiral members of a series of diastereoisomers in which at least one is chiral